

1 Introduction

Coupled differential equations describing metabolite concentration in culture media, cytosol and bacterial micro-compartment through time are presented.

2 Concentrations in time

2.1 Culture media

□ Vanillate concentration in culture media, assumed constant to analyze steady state.

$$\frac{d[V_m]}{dt} = 0$$

□ Protocatechuate concentration in culture media, assumed constant to analyze steady state.

$$\frac{d[P_m]}{dt} = 0$$

□ Catechol concentration in culture media, assumed constant to analyze steady state.

$$\frac{d[C_m]}{dt} = 0$$

□ Muconic acid concentration in culture media, assumed constant to analyze steady state.

$$\frac{d[M_m]}{dt} = 0$$

2.2 Cytosol

□ Vanillate concentration in interbacterial media

$$\frac{d[V_c]}{dt} = j_{b_{V_i}} \cdot ([V_b] - [V_c]) + j_{b_{[V_m]}} \cdot ([V_b] - [V_c]) - j_{m_V} \cdot ([V_c] - [V_m]) - \frac{K_{cat}^{ligM} \cdot [ligM]_c \cdot [V_c]}{[V_c] + K_m^{ligM}}$$

- $j_{b_{V_i}}$ is the flux constant of Vanillate introduction to BMC through shell porus. Sub-sub-sub index "m" stands for "media",
- K_{cat}^{ligM} and K_m^{ligM} are catalytic and Michaelis constant of $ligM$ enzyme, respectively.
- $[ligM]_c$ is $ligM$ enzyme concentration in cytosol.
- j_{m_V} is Vanillate flux constant through membrane, assumed equal for in and out fluxes.

Same analysis applies for $j_{b_{P_i}, C_i, M_i}$ \square Protocatechuate concentration in interbacterial media

$$\begin{aligned} \frac{d[P_c]}{dt} = & \cdot j_{b_{P_i}} \cdot ([P_b] - [P_c]) + \cdot j_{b_{P_m}} \cdot ([P_b] - [P_c]) - \cdot j_{m_P} \cdot ([P_c] - [P_m]) \\ & + \frac{K_{cat}^{ligM} \cdot [ligM]_c \cdot [V_c]}{[V_c] + K_m^{ligM}} - \frac{K_{cat}^{aroY} \cdot [aroY]_c \cdot [P_c]}{[P_c] + K_m^{aroY}} \end{aligned}$$

\square Catechol concentration in interbacterial media

$$\begin{aligned} \frac{d[C_c]}{dt} = & \cdot j_{b_{C_i}} \cdot ([C_b] - [C_c]) + \cdot j_{b_{C_m}} \cdot ([C_b] - [C_c]) - \cdot j_{m_C} \cdot ([C_c] - [C_m]) \\ & + \frac{(K_{cat}^{aroY} \cdot [aroY]_c \cdot [P_c])}{([P_c] + K_m^{aroY})} - \frac{K_{cat}^{catA} \cdot [catA]_c \cdot [C_c]}{[C_c] + K_m^{catA}} \end{aligned}$$

\square Muconic acid concentration in interbacterial media

$$\frac{d[M_c]}{dt} = j_{b_{M_i}} \cdot ([M_b] - [M_c]) + \cdot j_{b_{M_m}} \cdot ([M_b] - [M_c]) - \cdot j_{m_M} \cdot ([M_c] - [M_m]) + \frac{K_{cat}^{catA} \cdot [catA]_c \cdot [C_c]}{[C_c] + K_m^{catA}}$$

2.3 BMC

\square Vanillate concentration inside BMC.

$$\frac{d[V_b]}{dt} = j_{b_{V_i}} \cdot ([V_b] - [V_c]) - j_{b_{V_m}} \cdot ([V_b] - [V_c]) - \frac{K_{cat}^{ligM} \cdot [ligM]_b \cdot [V_b]}{[V_b] + K_m^{ligM}}$$

\square Protocatechuate concentration inside BMC.

$$\frac{d[P_b]}{dt} = j_{b_{P_i}} \cdot ([P_b] - [P_c]) - j_{b_{P_m}} \cdot ([P_b] - [P_c]) + \frac{K_{cat}^{ligM} \cdot [ligM]_b \cdot [V_b]}{[V_b] + K_m^{ligM}} - \frac{K_{cat}^{aroY} \cdot [aroY]_b \cdot [P_b]}{[P_b] + K_m^{aroY}}$$

\square Cathecol concentration inside BMC.

$$\frac{d[C_b]}{dt} = j_{b_{C_i}} \cdot ([C_b] - [C_c]) - j_{b_{C_o}} \cdot ([C_b] - [C_c]) + \frac{K_{cat}^{aroY} \cdot [aroY]_b \cdot [P_b]}{[P_b] + K_m^{aroY}} - \frac{K_{cat}^{catA} \cdot [catA]_b \cdot [C_b]}{[C_b] + K_m^{catA}}$$

\square Muconic acid concentration inside BMC.

$$\frac{d[M_b]}{dt} = j_{b_{M_i}} \cdot ([M_b] - [M_c]) - \cdot j_{b_{M_m}} \cdot ([M_b] - [M_c]) + \frac{K_{cat}^{catA} \cdot [catA]_b \cdot [C_b]}{[C_b] + K_m^{catA}}$$

2.4 BMC without ligM

We make a second model where ligM were not introduced to BMC, that is because tetrahydrofolate may not be able to get into a BMC for his big size.

□ Vanillate concentration inside BMC.

$$\frac{d[V_b]}{dt} = j_{b_{V_i}} \cdot ([V_b] - [V_c]) - j_{b_{V_m}} \cdot ([V_b] - [V_c])$$

□ Protocatechuate concentration inside BMC.

$$\frac{d[P_b]}{dt} = j_{b_{P_i}} \cdot ([P_b] - [P_c]) - j_{b_{P_m}} \cdot ([P_b] - [P_c]) - \frac{K_{cat}^{aroY} \cdot [aroY]_b \cdot [P_b]}{[P_b] + K_m^{aroY}}$$

□ Cathecol concentration inside BMC.

$$\frac{d[C_b]}{dt} = j_{b_{C_i}} \cdot ([C_b] - [C_c]) - j_{b_{C_o}} \cdot ([C_b] - [C_c]) + \frac{K_{cat}^{aroY} \cdot [aroY]_b \cdot [P_b]}{[P_b] + K_m^{aroY}} - \frac{K_{cat}^{catA} \cdot [catA]_b \cdot [C_b]}{[C_b] + K_m^{catA}}$$

□ Muconic acid concentration inside BMC.

$$\frac{d[M_b]}{dt} = j_{b_{M_i}} \cdot ([M_b] - [M_c]) - j_{b_{M_m}} \cdot ([M_b] - [M_c]) + \frac{K_{cat}^{catA} \cdot [catA]_b \cdot [C_b]}{[C_b] + K_m^{catA}}$$

2.5 Expression

□ Transcription rate from Meyer, A.J. et al. adjustment.

$$\frac{d[mRNA]}{dt} = \frac{c * 0.03 \frac{PoPS}{RPU} * \left(Ymin + (Ymax - Ymin) \frac{X^n}{K^n + X^n} \right)}{V} - \gamma_{mRNA} [mRNA]$$

□ ligM expression.

$$\frac{d[ligM]}{dt} = a [mRNA] - (\gamma_{ligM}) [ligM]$$

□ aroY expression.

$$\frac{d[aroY]}{dt} = a [mRNA] - (\gamma_{aroY}) [aroY]$$

□ catA expression.

$$\frac{d[catA]}{dt} = a [mRNA] - (\gamma_{catA}) [catA]$$

3 Integrating ODE's

We use python to integrate all this ODE's, all models can be found at: <https://github.com/uanligem/igem-2019>

3.1 Expression model

□ Parameters

$K1_{cat} = (0.078 \cdot 9.14) \cdot K_{cat} \text{ LigM}$
 $K1_m = 0.078 \text{ Km LigM}$
 $K2_{cat} = 0.00009 \text{ Kcat AroY}$
 $K2_m = 0.59 \cdot \text{Km AroY}$
 $K3_{cat} = (0.00185 \cdot 10485) \cdot K_{cat} \text{ CatA}$
 $K3_m = 0.00185 \text{ Kmm CatA}$
 $vol_o = 1000000 \text{ Medium volume (outside)}$
 $vol_c = 1e-15 \text{ Citoplasm volume}$
 $vol_b = 0 \text{ BMC volume}$
 $Y_m = 0.0043 \text{ mRNA degradation rate}$
 $n = 15 \text{ Number of plasmid copies}$
 $PoPS = 0.03 \text{ Transcription rate}$
 $Y_{min} = 2.4e-3 \text{ Min transcription rate (RPU's)}$
 $Y_{max} = 3 \text{ Max transcription rate (RPU's)}$
 $K = 26 \text{ Marionette constant (uM)}$
 $n2 = 2.3 \text{ Another marionette constant}$
 $a = 29.97 \text{ ligM traduction rate}$
 $b = 26.30 \text{ aroY traduction rate}$
 $c = 41.30 \text{ catA traduction rate}$
 $E_{cr} = 0.000385 \text{ ligM dilution rate}$
 $EE_{cr} = 0.000385 \text{ aroY dilution rate}$
 $EEE_{cr} = 0.000385 \text{ catA dilution rate}$
 $Vo = 3e-3 \text{ vanilic acid outside}$
 $Po = 0 \text{ protocatechuic acid outside}$
 $Co = 0 \text{ catechol outside}$
 $Mo = 0 \text{ cis,cis-muconic acid outside}$
 $\text{flux constants for species through membrane}$
 $j_{mV} = 1e - 13Va$
 $j_{mP} = 1e - 13PCA$
 $j_{mC} = 1e - 13Catechol$
 $j_{mM} = 1e - 13MA$

3.2 Modified BMC

□ Parameters $K1_{cat} = (0.078 \cdot 9.14) \text{ Kcat LigM}$

$K1_m = 0.078 \text{ Km LigM}$
 $K2_{cat} = 0.00009 \cdot 1 \text{ Kcat AroY}$
 $K2_m = 0.59 \text{ Km AroY}$
 $K3_{cat} = (0.00185 \cdot 10485) \text{ Kcat CatA}$
 $K3_m = 0.00185 \text{ Kmm CatA}$
 $vol_o = 1000000 \text{ Medium volume (outside)}$
 $vol_c = 1e-15 \text{ Citoplasm volume}$
 $vol_b = 0.1e-15 \text{ BMC volume}$

$V_o = 3e-3$ vanilic acid outside
 $P_o = 0$ protocatechuic acid outside
 $C_o = 0$ catechol outside
 $M_o = 0$ cis,cis-muconic acid outside
 $E_c = 0.1479$
 $EE_c = 0.04329$
 $EEE_c = 0.2039$
 $E_b = 1.2 * E_c$
 $EE_b = 1.2 * EE_c$
 $EEE_b = 1.2 * EEE_c$
 flux constants for species through membrane
 $j_{mV} = 1e - 13 V_{A\text{throughMembrane}}$
 $j_{mP} = 1e - 13 PCA$
 $j_{mC} = 1e - 13 Catechol$
 $j_{mM} = 1e - 13 MA$
 $j_{bVo} = 0.1e - 13 \text{through pduU}$
 $j_{bPo} = 0.1e - 13$
 $j_{bCo} = 0.1e - 13$
 $j_{bMo} = 0.1e - 13$
 $j_{bVi} = 1e - 15 pduA$
 $j_{bPi} = 1e - 15$
 $j_{bCi} = 1e - 15$
 $j_{bMi} = 1e - 15$

3.3 Modified BMC

□ Parameters $K1_{cat} = (0.078 * 9.14)$ Kcat LigM
 $K1_m = 0.078$ Km LigM
 $K2_{cat} = 0.00009 * 1$ Kcat AroY
 $K2_m = 0.59$ Km AroY
 $K3_{cat} = (0.00185 * 10485)$ Kcat CatA
 $K3_m = 0.00185$ Kmm CatA
 $vol_o = 1000000$ Medium volume (outside)
 $vol_c = 1e-15$ Citoplasm volume
 $vol_b = 0.1e-15$ BMC volume
 $V_o = 3e-3$ vanilic acid outside
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 $E_c = 0.1479$
 $EE_c = 0.04329$
 $EEE_c = 0.2039$
 $E_b = 1.2 * E_c$
 $EE_b = 1.2 * EE_c$
 $EEE_b = 1.2 * EEE_c$
 flux constants for species through membrane

$j_{mV} = 1e - 13$ *V AtroughMembrane*
 $j_{mP} = 1e - 13$ *PCA*
 $j_{mC} = 1e - 13$ *Catechol*
 $j_{mM} = 1e - 13$ *MA*
 $j_{bVo} = 0e - 13$ *throughpduU*
 $j_{bPo} = 0e - 13$
 $j_{bCo} = 0e - 13$
 $j_{bMo} = 0e - 13$
 $j_{bVi} = 1e - 15$ *pduA*
 $j_{bPi} = 1e - 15$
 $j_{bCi} = 1e - 15$
 $j_{bMi} = 1e - 15$